

REVIEW

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Biochemistry and biotechnology of mesophilic and thermophilic nitrile metabolizing enzymes

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Abstract Mesophilic nitrile-degrading enzymes are widely dispersed in the Bacteria and lower orders of the eukaryotic kingdom. Two distinct enzyme systems, a nitrilase catalyzing the direct conversion of nitriles to carboxylic acids and separate but cotranscribed nitrile hydratase and amidase activities, are now well known. Nitrile hydratases are metalloenzymes, incorporating Fe^{III} or Co^{II} ions in thiolate ligand networks where they function as Lewis acids. In comparison, nitrilases are thiol-enzymes and the two enzyme groups have little or no apparent sequence or structural homology. The hydratases typically exist as $\alpha\beta$ dimers or tetramers in which the α - and β -subunits are similar in size but otherwise unrelated. Nitrilases however, are usually found as homomultimers with as many as 16 subunits. Until recently, the two nitrile-degrading enzyme classes were clearly separated by functional differences, the nitrile hydratases being aliphatic substrate specific and lacking stereoselectivity, whereas the nitrilases are enantioselective and aromatic substrate specific. The recent discovery of novel enzymes in both classes (including thermophilic representatives) has blurred these functional distinctions. Purified mesophilic nitrile-degrading enzymes are typically thermolabile in buffered solution, rarely withstanding exposure to temperatures above 50°C without rapid inactivation. However, operational thermostability is often increased by addition of aliphatic acids or by use of immobilized whole cells. Low molecular stability has frequently been cited as a reason for the limited industrial application of “nitrilases”; such statements notwithstanding, these enzymes have been successfully applied for more than a decade to the kiloton production of acrylamide and more recently to the smaller-scale production of nicotinic acid, R-(–)-mandelic acid and

S-(+)-ibuprofen. There is also a rapidly growing catalog of other potentially useful conversions of complex nitriles in which the regioselectivity of the enzyme coupled with the ability to achieve high conversion efficiencies without detriment to other sensitive functionalities is a distinct process advantage.

Key words Mesophile · Thermophile · Nitrile hydratase · Nitrilase · Nitrile · Thermostable · Biotransformation

Distribution of nitrile-degrading enzyme systems

Nitrile (organocyanide) metabolizing enzyme systems are widely if relatively infrequently distributed through two of the three domains. The existence of nitrilase activity in only 3 of 21 plant families (Gramineae, Cruciferae, and Musaceae) (Thimann and Mahadevan 1964a) and in a limited number of fungal genera (e.g., *Fusarium*, *Aspergillus*, *Penicillium*) (Harper 1977a) indicates the relative rarity of this activity. Nitrile-degrading activities are apparently more frequent in Bacteria (see Tables 1 and 2), although it is impossible to assess the true distribution frequency accurately without access to extensive screening data. It is probable that entire groups of microorganisms have never been screened while others may have yielded negative and hence unreported results. A determination of nitrile-degrading activities in catalogs of thermophilic *Bacillus* and *Thermus* strains in the authors' laboratory (Cramp, Pereira, and Cowan, unpublished results) gave less than 10% and 0% hits respectively, while Duran reported only two positive signals in a Southern hybridization analysis of 31 bacterial species from 17 genera (Duran et al. 1993). To the author's knowledge, no reports currently exist of nitrile-degrading activity in the Archaea. Furthermore, none of the extant complete or partial archaeal genome sequences (Bult et al. 1996; Klenk et al. 1997; Sensen et al. 1996; Smith et al. 1997; Stein et al. 1996) show any significant homology to bacterial or eukaryotic nitrile hydratase or nitrilase gene sequences.

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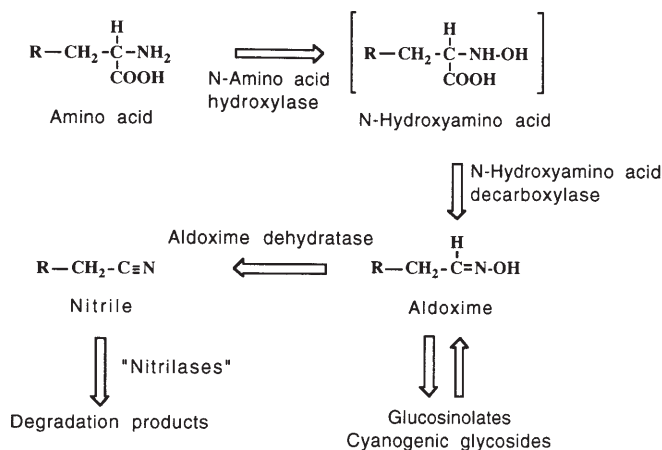


Fig. 1. Role of nitrile-degrading enzymes in plant nitrogen turnover

The apparent distribution of bacterial nitrile-degrading enzymes (e.g., *Rhodococcus*, *Nocardia*, *Brevibacterium*, *Pseudomonas*, and *Corynebacterium*; see Tables 1 and 2) may owe more to screening technologies than to molecular evolution or gene transfer. The most common screening protocols involve enrichment isolation in nitrile-supplemented C- or N-limited medium, inevitably skewing positive clones to selected groups of resilient and readily culturable microorganisms.

The physiological role of nitrile-degrading enzymes in bacteria is by no means clear. In plants, such activities are implicated in nutrient mobilization, particularly in the degradation of glucosinolates (Bestwick et al. 1993). It has also been suggested that nitrile-degrading enzymes form components of complex pathways controlling both the production and degradation of cyanogenic glycosides and related compounds and where aldoximes are key intermediates (Fig. 1). While such a role has yet to be established in bacterial systems, it is noted that some of the upstream enzyme activities, in particular the key enzyme aldoxime dehydratase, have been detected in *Pseudomonas* (Milborrow 1963).

Biochemical properties of mesophilic nitrile-degrading enzymes

The bacterial conversion of nitriles involves two well-established and quite separate enzyme systems (Fig. 2). The nitrilases, which frequently show a strong preference for aromatic substrates, convert nitriles directly to the corresponding acid. Alternatively, conversion via the amide intermediate involves two separately encoded but coordinately expressed enzyme activities, nitrile hydratase and amidase. The adjacent (and sometimes overlapping) positioning of the open reading frames encoding the nitrile hydratase α - and β -subunits and the amidase monomer suggest that these genes may be cotranscribed as a single mRNA (Duran et al. 1993; Nishiyama et al. 1991). Nitrilases and some nitrile hydratases are inducible proteins. Analysis

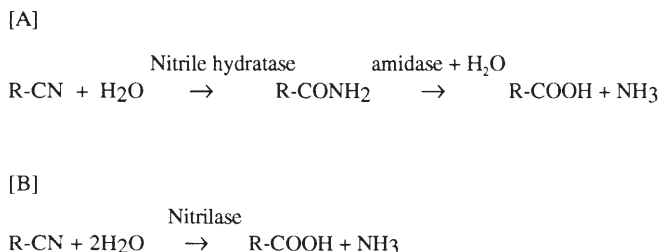


Fig. 2. Metabolism of nitriles: [A] nitrile hydratase/amidase; [B] Nitrilase

of a limited number of nitrilase and nitrile hydratase operations suggests that both upstream (Komeda et al. 1996a,b) and downstream (Hashimoto et al. 1994; Nishiyama et al. 1991) open reading frames are involved in positive transcriptional regulation.

Bacterial nitrile hydratases show considerable sequence and structural homology. For example, the α - and β -subunit sequence homologies identified between *Pseudomonas chlororaphis* B23 and *Rhodococcus* sp. N-774 were 63% and 58%, respectively (Ikehata et al. 1989; Nishiyama et al. 1991). Comparisons of shorter N-terminal amino acid sequences show high levels of sequence homology in the β -subunit (Fig. 3) but no significant homology in the α -subunit. The explanation for the latter has only become apparent with the publication of the first detailed X-ray crystallographic structure of a nitrile hydratase. Although the conserved N-terminal sequence of the β -subunit was buried in the center of the $\alpha\beta$ heterodimeric interface (Huang et al. 1997), the N-terminal sequence of the α -subunit was not buried, although still involved in $\alpha\beta$ dimer stabilization by wrapping across the surface of the β -subunit.

Alignments of nitrilase monomer amino acid sequences (Levy-Schil et al. 1995) from *Comamonas*, *Klebsiella*, *Alcaligenes*, and two *Rhodococcus* strains show only 13% identity, but with some residues conserved across eukaryotic nitrilase and cyanide hydratase sequences. One conserved cysteine has been implicated in catalytic activity by site-directed mutagenesis (Kobayashi et al. 1992a).

The native quaternary structures of bacterial nitrile hydratases are highly variable (Table 1), ranging from 30 to 505 kDa. However, with the exception of the homotetrameric *P. chlororaphis* enzyme (Nagasawa et al. 1987), all other known bacterial nitrile hydratases are $\alpha\beta$ heteromultimers, typically dimers or tetramers. There is also remarkable consistency in the sizes of the subunits: α , 24–28 kDa; β , 25–39 kDa. Nitrilases, by comparison, are homomultimers of approximately 38- to 76-kDa subunits with native compositions ranging from α_2 to α_{16} .

Both nitrile-degrading enzyme systems contain thiolate ligands at the active site. Inhibitor studies typically show sensitivity to sulfhydryl reagents, oxidizing agents, and heavy metals, but little or no reaction to serine inhibitors or, surprisingly, metal chelating agents. This result is inconsistent with the accepted view that most if not all nitrile

Table 1. Molecular and functional characteristics of mesophilic nitrile hydratases

Microorganism	Native MW (kDa)	Subunit MW (kDa)	Active center	T _{opt} (°C)	pH optimum	Substrate preference	Photoactivation	Production	Coding of structural genes	Reference
<i>Agrobacterium tumefaciens</i>		α 28	Fe ^{III} Co	25	6.5–9.5		ND	Inducible		Bauer et al. (1994)
<i>Arthrobacter</i> sp. J-1	420	α 24 β 27	-SH			Aliphatic	ND	Inducible		Asano et al. (1982)
<i>Corynebacterium nitrophilus</i>					6–8	Aliphatic, dinitriles, aromatic	ND			Amarant et al. (1989)
<i>C. pseudodiphtheriticum</i> ZBB-41	80	α 25 β 28	Fe ^{III}			Aliphatic	ND			Li et al. (1992)
<i>Corynebacterium</i> sp. C5	61	26.9	POQ? Fe ^{III}			Aliphatic, dinitriles, cyclic aliphatic	ND			Yamamoto et al. (1992b)
<i>Pseudomonas chlororaphis</i>	100	α 25 β 25	Fe ^{III}	20	6.0–7.5		–	Inducible	αβ	Nagasawa et al. (1987); Yamada et al. (1986)
<i>Pseudomonas putida</i>	54, 95	α 23 β 24.1	Co ^{II}							[Payne et al. (1997) #162]
<i>Rhodococcus equi</i> SHB-121	30	30				Aliphatic, cyclic	ND			Gilligan et al. (1993); Martinkova et al. (1995)
<i>R. erythropolis</i> BL1						Aliphatic, aromatic	ND	Inducible	αβ	Duran et al. (1993)
<i>R. rhodochrous</i> J1	505	α 26 β 29	Co	35–40	6.0–8.5	Aliphatic, aromatic	–	Inducible	βα	Kobayashi et al. (1991); Komeda et al. (1996c); Nagasawa et al. (1991a,b)
<i>Rhodococcus</i> sp. N-774 ^a	70	α 27 β 27	Fe ^{III}	35	7.0–8.5	Aliphatic	+	Constitutive	αβ	Endo and Watanabe (1989); Hashimoto et al. (1994)
<i>Rhodococcus</i> sp. N-771 ^a	70	α 27 β 27	Fe ^{III} , POQ			Aliphatic	+	Inducible	αβ	Ikehata et al. (1989); Yamada and Kobayashi (1996)
<i>Rhodococcus</i> sp. R312 ^{a,b}	95	α β	Fe ^{III}	35	7–8.5	Aliphatic, cyclic	+		αβ	Nagasawa et al. (1986)

PPQ, pyrroloquinoline quinone; ND, not determined.

^a Sequences identical.

^b Originally designated as *Brevibacterium* R312.

Fig. 3. Alignment of N-terminal amino acid sequences from mesophilic and thermophilic nitrile hydratase β-subunits. Data from: ¹[Kobayashi, 1991 #105], ²[Duran, 1992 #98], ³[Nagasawa, 1991 #118], ⁴[Ikehata, 1989 #136], ⁵[Payne, 1997 #162], ⁶[Yamaki, 1997 #166]. Alignments after [Huang, 1997 #164]

β Subunit

	Residue number																			
	1							10											20	
<i>R. rhodochrous</i> J1-L ¹	M	D	G	I	H	D	L	G	G	R	A	G	L	G	P	I	K	P	E	S
<i>R. rhodochrous</i> J1-H ¹	M	D	G	I	H	D	T	G	G	M	T	G	Y	G	P	V	P	Y	Q	K
<i>R. erythropolis</i> ²	M	D	G	V	H	D	L	A	G	V	Q	G	F	G	K	V	P	H	S	V
<i>P. chlororaphis</i> B23 ³	M	D	G	F	H	D	L	G	G	F	Q	G	F	G	K	V	P	H	T	I
<i>Rhodococcus</i> sp. N-774 ⁴	M	D	G	V	H	D	L	A	G	V	Q	G	F	G	K	V	P	H	T	V
<i>Rhodococcus</i> sp. R312 ³	M	D	G	V	H	D	L	A	G	V	Q	G	F	G	K	V	P	H	T	V
<i>Pseudomonas putida</i> ⁵	M	N	G	I	H	D	T	G	G	A	H	G	Y	G	P	V	Y	R	E	P
<i>Pseudonocardia thermophila</i> ⁶	M	N	G	V	Y	D	V	G	G	T	D	G	L	G	P	I	N	R	P	A
<i>Bacillus</i> sp. RAPc8	M	N	G	I	H	D	V	G	G	M	D	G	F	G	K	V	M	Y	V	K

Data from: ¹ [Kobayashi, 1991 #105], ² [Duran, 1992 #98], ³ [Nagasawa, 1991 #118], ⁴ [Ikehata, 1989 #136], ⁵ [Payne, 1997 #162], ⁶ [Yamaki, 1997 #166]. Alignments after [Huang, 1997 #164].

hydratases contain either low-spin nonheme Fe^{III} or noncorrinoid Co^{II} cofactors (Huang et al. 1997). Suggestions that *P. chlororaphis* B23 and *Brevibacterium* R312 nitrile hydratases contained pyrroloquinoline quinone (PQQ) prosthetic groups (Nagasawa and Yamada 1987; Sugiura et al. 1988) have more recently been discarded (see Kopf et al. 1996). The *Rhodococcus* R312 (*Brevibacterium* R312 has been reclassified as *Rhodococcus* R312) nitrile hydratase binds two atoms of Fe within the α -subunit, using three cysteine thiolates and two main chain amide nitrogen atoms as ligands. A subset of the low-spin nonheme iron-containing nitrile hydratases show an interesting photoactivation effect (Nagamune et al. 1990). Cells of *Rhodococcus* sp. N771 grown in the dark show no nitrile hydratase activity, but become catalytically active after light irradiation. Using Mossbauer, electron spin resonance (ESR), and magnetic susceptibility measurements (Honda et al. 1992), the oxidation of Fe^{II} to an Fe^{III} state has been

implicated in photoactivation. Most recently, it has been shown that the reduced Fe of the inactive enzyme binds NO (nitric oxide) in stoichiometric quantities (Bonnet et al. 1997) and that NO is released on photoactivation. The importance of NO as a neurotransmitter and physiological regulator in higher eukaryotes notwithstanding, the physiological significance of this process in *Rhodococcus* [and possibly in several other bacterial nitrile hydratases (Bonnet et al. 1997)] is not currently obvious.

The catalytic mechanism of iron-containing nitrile hydratases has been suggested to involve the Fe^{III} centre as a Lewis acid (Huang et al. 1997), either directly or indirectly activating the nitrile for hydration (Fig. 4). Mechanisms involving direct coordination of the nitrile nitrogen to the metal ion or attack on the nitrile by an iron-coordinated hydroxyl group are thought to be less likely, because trivalent metal ions typically show slow rates of ligand exchange (Huang et al. 1997). Indirect attack via a water molecule (as shown in Fig. 4) does not require ligand exchange and is consistent with the role of the Fe^{III} as a Lewis acid. Nucleophilic attack on the nitrile by an active site thiolate as proposed by Kopf et al. (1996) is probably not significant in *Rhodococcus* R312 because the crystal structure shows all thiolate ligands to be strongly coordinated to the Fe center (Huang et al. 1997).

The catalytic mechanisms of the nitrilases are less well established. A single conserved cysteine residue in a number of nitrilase sequences (Levy-Schil et al. 1995) and inhibitor studies (Kobayashi et al. 1989) are consistent with

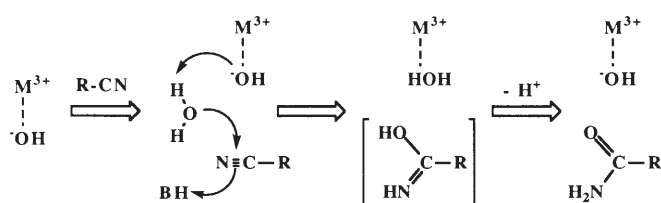


Fig. 4. Proposed mechanism for nitrile hydratase catalysis. (Redrawn from Huang et al. 1997)

Table 2. Molecular and functional characteristics of mesophilic nitrilases

Microorganism	Native and (subunit) MW (kDa)	Optimum pH	T_{opt} ($^{\circ}\text{C}$)	Substrate specificity	pI	Inducible	Gene cloned	Other characteristic	References
<i>Acinetobacter</i> sp. AK226	580 (43)	8	50	Aliphatic, aromatic, heterocyclic	3.99	No	?	Enantioselective	Yamamoto and Komatsu (1991)
<i>Arthrobacter</i> sp. strain J-1									
A	30	8.5	40	Aromatic and heterocyclic nitriles	4.95	Yes	?	Nitrile hydratase and amidase activity	Bandyopadhyay et al. (1986)
B	23	7.5	30	Aromatic and heterocyclic nitriles	4.8	Yes	?		
<i>Alcaligenes faecalis</i> JM3	275 (44)	7.5	45	Arylacetonitriles	4.9	Yes	Yes	Arylacetonitrilase	Kobayashi et al. (1993); Nagasawa et al. (1990)
<i>A. faecalis</i> ATCC 8750	460 (32)	7.5	40–45	Substituted aliphatic nitriles	?	Yes	?	Arylacetonitrilase and enantioselective activities	Yamamoto et al. (1992a)
<i>Comamonas testosteroni</i>	(38.7)			Dinitriles			Yes	High activity on adiponitrile	Levy-Schil et al. (1995)
<i>Fusarium solani</i>	620 (76)	7.8–9.1		Aromatic and heterocyclic nitriles	4.19	Yes	?		Harper (1977a)
<i>F. oxysporum</i> f. sp. <i>melonis</i>	550 (37)	6.0–11	40	Aromatic and heterocyclic nitriles	<4	Yes	?		Goldlust and Bohak (1989)
<i>Nocardia</i> sp. NCIB 11216	560 (45)	8		Aromatic nitriles	4.22	Yes	?		Harper (1977b)
<i>Nocardia</i> sp. NCIB 11215	560 (46)	7.0–7.5		Aromatic and heterocyclic nitriles	4.35	Yes	?		Harper (1985)
<i>Rhodococcus rhodochrous</i> J1	78 (41.5)	7.6	45	Aromatic and heterocyclic nitriles	5.6	Yes	Yes	Regiospecific	Kobayashi et al. (1992a, 1989)
<i>Rhodococcus</i> ATCC 39484	560 (40)	7.5	40	Aromatic and heterocyclic nitriles	?	Yes	?	Enantioselective	Stevenson et al. (1992)
<i>R. rhodochrous</i> K22	650 (41)	5.5	50	Aliphatic nitriles	?	Yes	Yes		Kobayashi et al. (1992b)

Table 3. Substrate specificity of *Bacillus pallidus* DAC521 nitrilase

Substrate	Relative activity (%)
Aliphatic	
Saturated	
Acetonitrile	3
Chloroacetonitrile	10
Trichloroacetonitrile	0
Propionitrile	0
Butyronitrile	19
Isobutyronitrile	0
4-Chlorobutyronitrile	110
Valeronitrile	43
Isovaleronitrile	0
Saturated dinitrile	
Malononitrile	0
Succinonitrile	0
Glutaronitrile	11
Adiponitrile	14
Iminodiacetonitrile	0
3,3-Iminodipropionitrile	0
Unsaturated	
Acrylonitrile	9
Methacrylonitrile	2
Allylcyanoide	20
Crotononitrile	80
cis-2-pentenitrile	0
trans-3-pentenitrile	23
3-Aminopropionitrile fumarate	0
Unsaturated dinitrile	
Fumaronitrile	16
Mucononitrile	2
Heterocyclic	
2-Cyanopyridine	92
3-Cyanopyridine	135
4-Cyanopyridine	233
1-Cyclopentene acetonitrile	0
1-Cyanoacetylpiperidine	0
Aromatic	
Benzonitrile	100
2-Aminobenzonitrile	0
3-Aminobenzonitrile	24
4-Aminobenzonitrile	3
2-Chlorobenzonitrile	3
3-Chlorobenzonitrile	50
4-Chlorobenzonitrile	192
2-Hydroxybenzonitrile	0
3-Hydroxybenzonitrile	57
4-Hydroxybenzonitrile	29
2-Fluorobenzonitrile	27
3-Fluorobenzonitrile	157
4-Fluorobenzonitrile	156
2-Nitrobenzonitrile	2
3-Nitrobenzonitrile	21
4-Nitrobenzonitrile	23
<i>o</i> -Tolunitrile	44
<i>m</i> -Tolunitrile	30
<i>p</i> -Tolunitrile	77
3-Chloro-4-fluorobenzonitrile	95
3,4-Dibromo-4- <i>OH</i> -benzonitrile	0
Phenylacetoneitrile	1
4-Aminobenzylcyanide	0
Mandelonitrile	0
2-(Methoxyphenyl)acetoneitrile	6
3-(Benzylamino)propionitrile	0
Cinnamonitrile	6
Benzylidenemalononitrile	0

The relative activity of the compounds shown was measured at 50°C and pH 7.0 with a substrate concentration of 100 mM in the presence of 100 mM urea. Relative activities are expressed relative to benzonitrile (100%).

nucleophilic attack on the nitrile carbon by an activated thiol residue (Harper 1977; Thimann and Mahadevan 1964b). Metal ions have not been implicated in the catalytic process. The role of an active site thiolate nucleophile is further supported by the recent report of the generation of nitrilase activity in the cysteine protease papain by a Gln19Glu mutation in the active site (Dufour et al. 1995).

Tables 1 and 2 show the general functional properties of bacterial nitrile hydratases and nitrilases. The relatively low temperature optima of both groups of enzymes reflect the growth optima of the source organisms and the intracellular nature of the enzymes. In some cases, these values include in vitro stabilization by compounds such as alkanoic acids (Nagasawa et al. 1987). Nevertheless, there is a consensus that the Co-containing hydratases are generally more thermostable than the iron-containing enzymes (Payne et al. 1997). The neutral-to-alkaline pH optima are probably as much influenced by the acid-dependent hydrolysis of organocyanides as by pH-dependent changes in active site residue (thiolate) ionization.

The specificities of mesophilic nitrile hydratases and nitrilases are only cursorily shown in Tables 1 and 2, the reader being directed to several excellent reviews (Meth-Cohn and Wang 1997a,b; Sugai et al. 1997a,b) and to the original papers for more detailed information. What is clear, however, is that the early dogma that “nitrile hydratases are aliphatic substrate specific and lack enantioselectivity” while “nitrilases are aromatic substrate specific and enantioselective” has been severely shaken by the discovery of nitrile hydratases capable of catalyzing the hydration of linear, branched, and cyclic aliphatic nitriles together with substituted and complex aromatic nitriles (Meth-Cohn and Wang 1997a,b), by the demonstration of stereoselectivity in purified nitrile hydratases (as opposed to whole-cell systems where the presence of the amidase may complicate interpretation) (Martinkova et al. 1996) and by the characterization of very broad specificity nitrilases (Almatawah and Cowan, unpublished results) (Table 3).

Biotransformations of nitriles

Probably the major driving force behind studies of nitrile-metabolizing enzymes comes from the potential capacity of these enzymes to perform valuable biotransformations. Nitriles are important intermediates in chemical synthesis, particularly as products of one-, two-, and three-carbon additions to core synthons. Although the chemical conversion of nitriles to amides and acids is facile, relatively harsh reactions conditions (heat, acid, or alkali) are required. Such conditions may be incompatible with the presence of sensitive functionalities such as ester-linked groups. The ability to convert nitriles to amides or acids under very mild conditions (neutral to mildly alkaline pH, ambient temperatures), especially with the advantage of chemo-, regio-, or stereoselectivity, underpins the growing interest in nitrile-metabolizing enzymes.

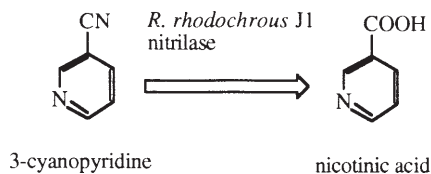


Fig. 5. Biosynthesis of nicotinic acid

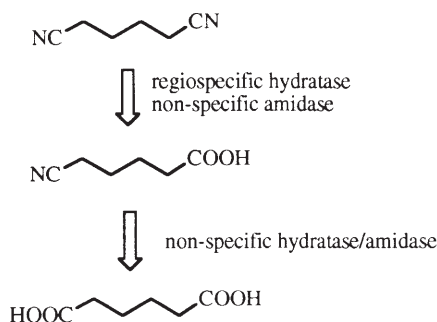


Fig. 6. Conversion of adiponitrile to adipic acid

In one aspect, the biotechnology of nitrile biotransformations has already matured. The success of three generations of nitrile-metabolizing bacteria, *Rhodococcus* N-774, *Pseudomonas chlororaphis* B23, and *R. rhodochrous* J1, in the kiloton production of acrylamide (Nagasawa et al. 1993; Yamada and Kobayashi 1996) has unarguably demonstrated the commercial viability of these enzymes. The biosynthesis of nicotinic acid from 3-cyanopyridine (Mathew et al. 1988) (Fig. 5) is also on the verge of large-scale commercialization.

This review is not an appropriate forum for a detailed summary of nitrile biotransformations, but merely attempts to demonstrate the potential scope of “nitrilase” applications by reviewing recent examples.

Example 1: synthesis of cyanovaleric acid and adipic acid

The conversion of adiponitrile to cyanovaleric acid and hence to adipic acid (Fig. 6) provides a synthetic route to the production of intermediates in the formation of the polymers nylon 6 and nylon 6,6. Nitrilases showing high activity on adiponitrile and the cyano-acid intermediate have only recently been identified (Moreau et al. 1994). For example, a *Comamonas testosteroni* nitrilase was reported (Levy-Schil et al. 1995) to have a specific activity for conversion of adiponitrile to cyanovaleric acid (CVA) of 29 μ moles CVA produced/min/mg protein, and activity for the conversion of CVA to adipic acid of approximately 20% of this value. Relative activities on standard aliphatic (propionitrile) and aromatic (benzonitrile) substrates were substantially lower.

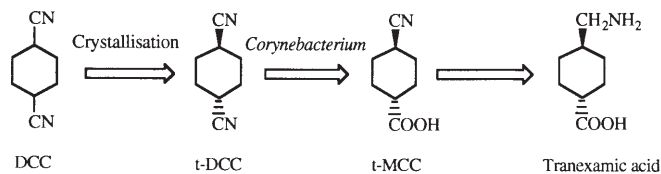


Fig. 7. Bacterial semisynthesis of tranexamic acid

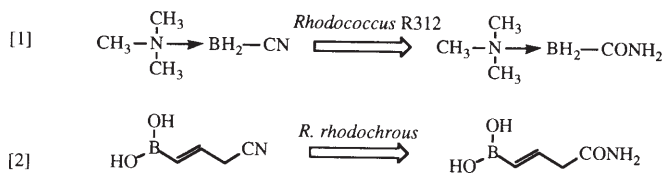


Fig. 8. Biotransformations of boron-containing nitriles

Example 2: semisynthesis of tranexamic acid

Tranexamic acid (*trans*-4-aminomethylcyclohexane-1-carboxylic acid) is an antifibrinolytic agent that is synthesized chemically from *cis*, *trans*-cyclohexane-1,4-dimethylcarboxylate (Fig. 7) and obtained as a pure *trans* product by repeated and difficult crystallization from a mixture of *cis* and *trans* products. A simpler separation of the *cis*, *trans*-dicyanocyclohexanes is precluded because of the lack of mononitrile specificity in the subsequent alkaline hydrolysis and conversion of the *trans* to the *cis* isomer. An alternative process for the biological conversion of *trans*-dicyanocyclohexane (t-DCC) to *trans*-4-cyanocyclohexane-1-carboxylic acid (t-MCC) (Yamamoto et al. 1992b) therefore avoids both problems associated with the chemical conversion. Reaction of insoluble t-DCC in aqueous solution (up to 60% by weight) at 4°C with resting cells of *Corynebacterium* C5 gave yields in excess of 99% with no transversion of stereochemical configuration.

Example 3: biotransformation of boron-containing nitriles

The production of boron-containing organic compounds offers a variety of synthetic routes to the generation of new and potentially valuable compounds. For example, the boron-containing amino acid analogs (amine-carboxyboranes) possess a variety of biological activities and have potential application as antineoplastic, antihyperlipidemic, analgesic, and antiarthritic agents (Spielvogel 1988). Such compounds may be chemically synthesized at low yield from amine cyanoboranes in an inert gas atmosphere, but the possibility of bioconversion using nitrilases potentially offers significant advantages (Fig. 8, scheme [1]).

Table 4. Structural and functional properties of thermophilic nitrile-metabolizing enzymes

Activity Source	Nitrile hydratase <i>Bacillus</i> sp.	Nitrile hydratase <i>B. pallidus</i>	Nitrile hydratase <i>Pseudonocardia thermophila</i>	Nitrilase <i>B. pallidus</i>
Growth T_{opt}	65°C	50°C	55°C	60°C
Enzyme composition	$\alpha_2\beta_2$	$\alpha_2\beta_2$	$\alpha_n\beta_m$	α_{14}
Native and subunit MW (kDa)	110 28, 29	110 27, 29	Unknown 29, 32	600 42.6
Specificity	Very broad: Linear, branched and cyclic aliphatics	Narrow: aliphatic	Unknown	Very broad: aromatic, heterocyclic, branched and cyclic aliphatic
Reference	Pereira et al. (this issue)	Cramp (1998); unpublished	Yamaki et al. (1997)	Almatawah et al., unpublished results

A recent preliminary report (Millais et al. 1997) demonstrating that *Rhodococcus* R312 is probably capable of catalyzing the slow conversion of trimethylamine-cyanoborane to the amide derivative trimethylamine-carboamidoborane suggests that such unusual reactions are worthy of further attention. Similarly, *R. rhodochrous* cells have been shown to rapidly convert nitrile-derivatized alkenylboronic acids to the corresponding amides (Fig. 8, scheme [2]) (Sugai et al. 1997a). Further transformation by intracellular amidase to the acid was slow and relatively low yields were obtained, even after prolonged incubation. Reaction of the amide intermediates with haloalkenes provides a route to the synthesis of derivatized alkenes with conjugated double bonds having defined configurations.

Thermophilic nitrile-metabolizing enzymes

Relatively few examples of thermophilic nitrile-metabolizing enzymes have yet been reported. Detailed publications are limited to nitrile hydratases derived from several moderately thermophilic *Bacillus* species (Cramp et al. 1997; Pereira et al., [this issue]; Takashima et al. 1995, 1996) and a new thermophilic actinomycete, *Pseudonocardia thermophila* (Yamaki et al. 1997). To the author's knowledge, no reports of thermophilic nitrilases have yet been published although one such enzyme has been characterized in detail in the author's laboratory (Almatawah and Cowan, unpublished data).

Some molecular and functional properties of the known thermophilic nitrile-metabolizing enzymes are shown in Table 4. The striking conclusion from these data is that, thermostability apart, the general structural and functional characteristics of these enzymes are remarkably similar to their mesophilic "homologs." This similarity is also evident in the sequence identity of the N-terminal regions of the nitrilases (Almatawah and Cowan, unpublished results) and nitrile hydratase β -subunits (see Fig. 3). This conclusion is consistent with the accepted view that thermophilicity in all Bacteria other than the few deeply branched genera such as the hyperthermophiles *Thermotoga* and *Aquifex* is a relatively recent evolutionary product.

Table 5. Thermal stability values of purified and crude (values in brackets) thermophilic nitrile-metabolizing enzymes

Half-life values at temperature in °C	<i>Bacillus pallidus</i> NHase	<i>Bacillus</i> strain NHase	<i>B. pallidus</i> Nase
30	7.0h (120h)	80h	– (7d)
37	– (67h)	45h	– (3.5d)
50	– (4.5h)	2.5h	8h (6.5h)
55	51m –	–	–
60	6.8m (8.2m)	<20min	2.5h (34m)
70	–	–	<10min –
80	–	–	<5min –

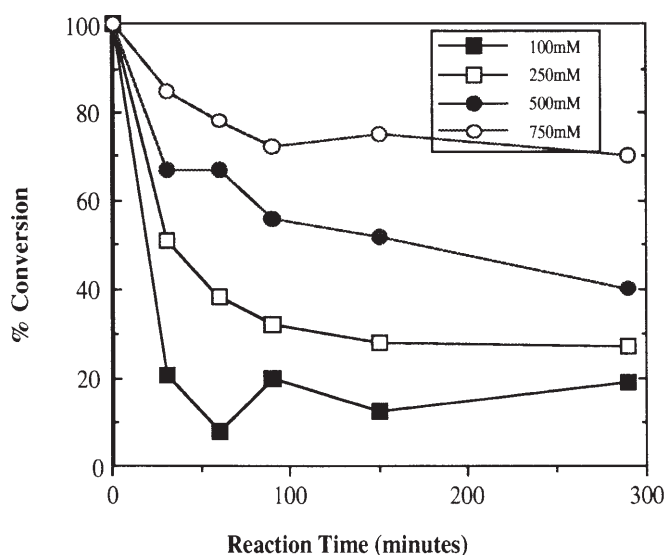
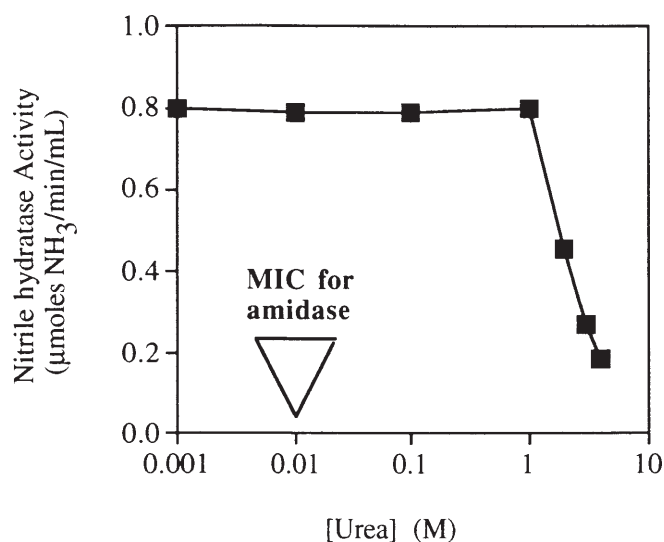
The thermostabilities of the thermophile-derived nitrile-metabolizing enzymes (Table 5) are all substantially higher than those observed for homologous enzymes derived from mesophilic organisms. As might be expected, the purified enzymes all show significant stability at temperatures around the optimal growth temperature of the source organisms. The stability of *B. pallidus* nitrile hydratase activity is, however, substantially greater in unbroken cells, indicating that the intracellular environment provided a significant level of extrinsic stabilization.

Many of the mesophilic nitrile-metabolizing enzymes are remarkably unstable, having very short activity half-lives in the growth temperature range of 20°–35°C [e.g., *Corynebacterium pseudodiphtheriticum*, $t_{1/2}$ = 65 min at 20°C (Li et al. 1992); *Pseudomonas chlororaphis* B23, $t_{1/2}$ = 11 min at 30°C (Nagasawa et al. 1987); *Rhodococcus rhodochrous* J1, $t_{1/2}$ = 58 min at 60°C (Nagasawa et al. 1991a,b); *Corynebacterium* sp. C5, $t_{1/2}$ = 7.5 min at 45°C (Tani et al. 1989)]. Furthermore, the values quoted for the *R. rhodochrous* and *Corynebacterium* sp. C5 enzymes are for stabilities in the presence of aliphatic acids (butyrate or valerate), compounds that strongly stabilize some mesophilic nitrile hydratases (Kopf et al. 1996). The relatively high stabilities of the thermophilic nitrile-metabolizing enzymes, their rapid rates of turnover even at suboptimal temperatures (Cramp et al. 1997; Pereira et al., [this issue]) and their broad substrate specificities (Table 6) all contribute to the view that these enzymes may in the future be successfully

Table 6. Comparative specificities of thermophilic *Bacillus* nitrile-metabolizing enzymes

Substrate	Structure	<i>B. pallidus</i> NHase	<i>Bacillus</i> RAPc8 NHase	<i>B. pallidus</i> Nase
Acetonitrile	CH ₃ CN	93	60	3
Propionitrile	CH ₃ CH ₂ CN	30	40	0
Butyronitrile	CH ₃ (CH ₂) ₂ CN	64	55	19
Valeronitrile	CH ₃ (CH ₂) ₃ CN	100	49	43
Acrylonitrile	CH ₂ =CHCN	62	79	9
Crotonitrile	CH ₃ CH=CHCN	38	100	80
Benzonitrile	C ₆ H ₅ CN	0	0	100
Cyclopentene-acetonitrile	C ₅ H ₇ CH ₂ CN	58	0	0
Malononitrile	NCCH ₂ CN	6	0	0
Glutaronitrile	NC(CH ₂) ₃ CN	30	0	11
Adiponitrile	NC(CH ₂) ₄ CN	27	0	14

Values are relative activities, normalized to the maximum value.

**Fig. 9.** Biotransformation of acrylonitrile with encapsulated thermophilic *Bacillus* cells**Fig. 10.** Effect of urea on *Bacillus* RAPc8 nitrile hydratase and amidase activities

applied to industrial biocatalysis and biotransformation processes.

Acrylamide production using thermophilic nitrile hydratases

Acrylonitrile hydrolysis trials using *Bacillus* RAPc8 cells immobilized by encapsulation in sodium alginate (Graham, Barfield, Pereira, and Cowan, manuscript in preparation) indicated that the intracellular hydratase and amidase activities were remarkably resistant to inactivation (Fig. 9), even at acrylonitrile concentrations approaching those required for efficient industrial biocatalysis (0.75–1.0M). In a

nonoptimized batch reaction, up to 500mM acrylonitrile substrate concentrations were efficiently catalyzed (greater than 60% conversion in 5 h). This *Bacillus* RAPc8-catalyzed whole-cell reaction converts acrylonitrile to acrylic acid via the amide, although high levels of intracellular amidase activity in this organism prevent accumulation of the amide intermediate (Pereira, unpublished results). Methods of effectively blocking amidase activity to produce the intermediate would obviously be of substantial value in the possible application of this organism. A screen of potential amidase inhibitors indicated that urea was particularly effective (MIC = 10 mM; 100% inhibition at <100 mM). The absence of any inhibitory effect of urea on nitrile hydratase activity at concentrations less than 1 M (Fig. 10) provides a substantial “operational window” in which the amide may be generated.

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